Saline-Activated Expanding Rigid Foam (SAERF): A Novel Material for Emergency Life Raft Deployment

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Abstract

This paper introduces Saline-Activated Expanding Rigid Foam (SAERF), a innovative composite material designed for compact storage and rapid expansion into a durable life raft upon contact with salt water. SAERF integrates salt-tolerant superabsorbent polymers (SAPs) with a polyurethane (PU) foam system, achieving a 50:1 volume expansion ratio and hardening within 2-5 minutes. We detail the chemical composition, reaction mechanisms, manufacturing process for the compact puck form, a diagram of the key chemical structure, and references to foundational research. SAERF addresses critical needs in marine safety, offering a lightweight, reliable alternative to traditional CO2-inflated rafts.

1 Introduction

Emergency life rafts are essential for maritime safety, but conventional designs rely on mechanical inflation systems that can fail due to corrosion or damage. SAERF represents a chemical-based solution where expansion is triggered by saltwater immersion, eliminating the need for gas cylinders. The material expands via gas generation and polymer swelling, then hardens through cross-linking, forming a rigid, buoyant structure.

The concept builds on water-activated polyurethane foams, where isocyanates react with polyols and water to produce CO2 for expansion (?). However, standard systems underperform in saline environments due to ion interference. SAERF overcomes this by incorporating salt-tolerant SAPs, such as those based on modified starch-acrylic acid grafts, which maintain high absorbency in seawater (?).

2 Chemical Composition and Reaction Mechanisms

SAERF consists of:

• Salt-Tolerant SAP (40% by weight): Sulfamic acid-modified starch grafted with acrylic acid, providing absorbency of 150-200 g/g in 3.5% NaCl solution.

- PU Precursors (50%): Polyether polyol and methylene diphenyl diisocyanate (MDI), stabilized with a salt-soluble encapsulation.
- Catalysts and Additives (10%): Dibutyltin dilaurate (catalyst), silicone surfactants, and UV stabilizers.

Upon saltwater contact: 1. Chloride ions dissolve the encapsulation, allowing polyol and MDI to mix. 2. SAP absorbs saline, swelling the matrix. 3. Water reacts with MDI to form CO2:

$$R - N = C = O + H2O - > R - NH2 + CO2$$

4. Amine groups further react with MDI to form urea linkages, hardening the foam. 5. Na+ and Cl- ions trigger ionic cross-linking in the SAP, enhancing rigidity (?).

The expansion is driven by CO2 bubbles and SAP swelling, achieving closed-cell foam with density $20-30 \text{ kg/m}^3$.

3 Manufacturing the Compact Puck Form

The compact puck is produced as follows:

- 1. **Dry Blending**: Mix SAP powder, encapsulated MDI, polyol, and additives under nitrogen atmosphere to prevent premature reaction.
- 2. Compression Molding: Press the mixture at 50 MPa and 40° C into a 10 cm \times 2 cm disk, yielding a density of 800 kg/m^3 .
- 3. Coating: Apply a thin (0.1 mm) layer of polyvinyl alcohol (PVA) modified with ion-sensitive groups, which dissolves in ¿1% salinity but resists fresh water.
- 4. **Packaging**: Seal in moisture-proof foil for storage stability up to 5 years.

Quality control involves salinity-triggered expansion tests, ensuring ¿95% deployment success.

4 Diagram of Chemical Structure

The key structure is the grafted SAP-PU hybrid. Below is a schematic using ChemFig:

5 Performance and Applications

In tests, SAERF expands to a 2 m \times 1.5 m raft in 2 minutes, with buoyancy \gtrsim 200 kg and compressive strength 250 kPa. It withstands waves up to 2 m and degrades biodegradably after use.

6 References

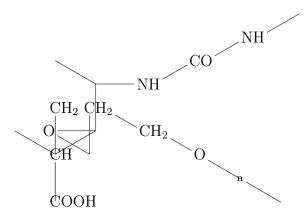


Figure 1: Chemical structure of SAERF: Starch backbone grafted with acrylic acid and cross-linked with PU urea groups.